661/4Chem.

UG/5th Sem/CHEM-H-DSE-T-2C/20

U.G. 5th Semester Examination - 2020

CHEMISTRY

[HONOURS]

Discipline Specific Elective (DSE)
Course Code: CHEM-H-DSE-T-2C

Full Marks : 40 Time : $2\frac{1}{2}$ Hours

The figures in the right-hand margin indicate marks.

Candidates are required to give their answers in their own words as far as practicable.

- 1. Answer any **five** questions: $2 \times 5 = 10$
 - i) What is the working definition of green chemistry?
 - ii) Why microwave is a non-ionizing radiation? Give an example of ionizing radiation.
 - iii) What are differences between microwave heating and conventional heating?
 - iv) What is supercritical fluid? Name few commonly used supercritical fluids.
 - v) Explain why biofuel is renewable source of energy but not fossil fuel?
 - vi) What are oils and fats? What is the difference between them?

vii) What is biodegradable polymer? Give an example that is industrially useful.

2. Answer any **five** questions:

 $3 \times 5 = 15$

- i) What is susceptors in microwave induced reaction? Give an example where ionic liquid acts as a susceptors. $1\frac{1}{2}+1\frac{1}{2}$
- ii) What is cavitation? Name different types of cavitation and specify which type is mainly responsible for chemical reaction. 1+2
- iii) Give the examples of microwave assisted oxidation and decarboxylation reactions.

$$1\frac{1}{2}+1\frac{1}{2}$$

iv) Predict the products A and B with suitable explanation.

$$A \xleftarrow{\frac{PhCH_3/Al_2O_3}{Sonication}} + KCN \xrightarrow{\frac{PhCH_3/Al_2O_3}{CH_3}} B$$

v) What is biodiesel? How does the use of side product during biodiesel synthesis reduce the cost of production? 1+2

- vi) Describe the role of supercritical CO₂ in dry cleaning industry.
- vii) Define 'Atom Economy', 'E-Factor' and 'Environmental Quotient.' 1+1+1
- 3. Answer any **three** questions: $5 \times 3 = 15$
 - i) Why oxidation of cyclohexane or cyclohexanol to adipic acid is not considered as green synthesis? Write down one green approach for the synthesis of adipic acid. 2+3
 - ii) Write down twelve principles of green chemistry.5
 - iii) Calculate the atom economy (AE) for the following reactions and explain which one is greener synthesis of propelene glycol and why?

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Reaction 1:

$$CH_{3}CH = CH_{2} \xrightarrow{Cl_{2}/H_{2}O} CH_{3}CHOHCH_{2}Cl \xrightarrow{NaOH} \bigcirc$$

$$\xrightarrow{\text{H}_2\text{O}, 120-190 °C} \text{CH}_3\text{CHOHCH}_2\text{OH}$$

Reaction 2:

$$\label{eq:hoch2} \begin{split} \text{HOCH}_2\text{CHOHCH}_2\text{OH} + \text{H}_2 & \xrightarrow{\text{CuCrO}_4, \text{ 200 °C}} \text{CH}_3\text{CHOHCH}_2\text{OH} \end{split}$$

- iv) a) What is ionic liquid? Why it is called designer solvent?
 - b) What characteristics of water make it benign solvent? 1+2+2