584 Chem/PR

UG/5th Sem/Chem(H)CC-12/PR/21

U.G. 5th Semester Examination - 2021 CHEMISTRY

[HONOURS]

Course Code : CHEM(H)CC-P-12/PR
[PRACTICAL]

Full Marks: 20 Time: 2 Hours

The figures in the right-hand margin indicate marks.

Candidates are required to given their answer in their own words as far as practicable.

Answer any **four** questions from the following: $5 \times 4 = 20$

- 1. Define molar extinction coefficient. What is the unit of molar extinction coefficient? How will you test the validity of Lambert-Beer's Law for K₂Cr₂O₇ solution? 1+1+3
- 2. Distinguish between surface-tension and surface energy. What is unit of surface-tension? What is the basis of determining surface-tension of a liquid using a stalagmometer? 1+1+3
- 3. What is the Critical Micelle Concentration (CMC) of surfactants? How is the CMC of a surfactant determined spectrophotometrically? 1+4

- 4. What is the unit of second order rate constant? Briefly discuss the principle for the spectrophotometrical study of the kinetics of the reaction between $S_2O_8^{2-}$ and I^- . 1+4
- 5. What are acid-base indicators? Give examples. How is the pH of an unknown buffer measured spectrophotometrically? 1+1+3
